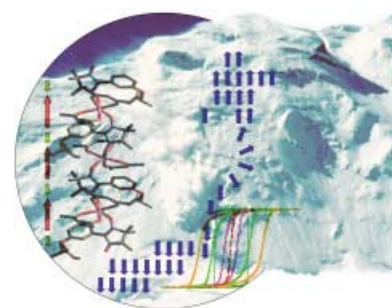


COVER PICTURE

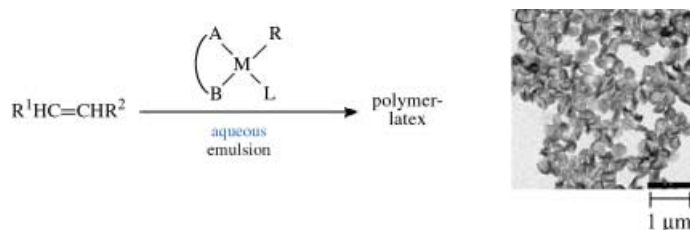
The cover picture shows a snow avalanche on the north face of Everest behind the structure of an enantiopure molecular ferromagnet which displays a rare phenomenon: a “magnetic avalanche”. The chiral complex $[\text{Mn}(\text{hfac})_2]$ ($S = \frac{1}{2}$, spins represented by the red arrows) with an enantiopure nitronyl nitroxide ligand ($S = \frac{1}{2}$, green arrows) exists as a coordination polymer (hfac = hexafluoroacetylacetonate). These chains are ferromagnetic (and are represented by the blue arrows), since the antiferromagnetically coupled organic- and metallic-centered spins do not compensate. At temperatures below 3 K these chains order magnetically and hysteresis loops open up. However, at 0.13 K (green curve) the width of the loop is smaller than at 0.3 K (orange curve). Thus, at a certain value of applied field there is an abrupt switching of some spins, which releases energy as a local heating. This heat initiates a switching of neighboring spins, which creates more heat and results in the magnetic equivalent of an avalanche. Further details about this optically active material are described by Veciana et al. on p. 586 ff. (Snow avalanche picture courtesy of Albert Castellet[®], and thanks to Pere Oller of the Institut Cartogràfic de Catalunya)



REVIEW

Contents

Water—a highly attractive, environmentally friendly reaction medium particularly for polymerization reactions. Whereas free-radical routes are well established, aqueous catalytic polymerization has seen major advances only recently. Numerous olefinic monomers can be transformed, for example, to aqueous polymer latices (see schematic reaction and picture).



S. Mecking,* A. Held,
F. M. Bauers 544–561

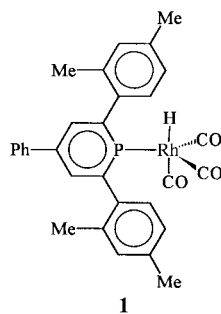
Aqueous Catalytic Polymerization of
Olefins

Keywords: colloids • homogeneous
catalysis • polymerizations • polymer
latex • water chemistry

Angew. Chem. **2002**, *114*, 564–582

Molecules with low-coordinate phosphorus atoms were considered as a domain for basic researchers for three decades. They were studied as building blocks in element–organic synthesis or as ligands in coordination chemistry. In the last few years transition metal complexes with phosphabenzene and phospholyl ligands have attracted attention as components in efficient catalysts. Thus, **1** is an efficient hydroformylation catalyst. This account also covers the use of such complexes for C–C coupling reactions, polymerization, and asymmetric synthesis.

Angew. Chem. **2002**, *114*, 583–592



L. Weber* 563–572

Phosphorus Heterocycles: From Laboratory Curiosities to Ligands in Highly Efficient Catalysts

Keywords: asymmetric catalysis • homogeneous catalysis • hydroformylation • phosphaheterocycles • P ligands

HIGHLIGHT

More than 200 years after the discovery of the element oxygen, recent experimental studies at very large as well as extremely low pressures suggest the existence of molecular tetraoxygen O₄ as a possible new modification of oxygen.

Angew. Chem. **2002**, *114*, 593–594

O₂, O₃, O₄, ... O_n?

D. Schröder* 573–574

News about Oxygen

Keywords: allotropes • atmospheric chemistry • mass spectrometry • oxygen

VIPs

The following communications are “Very Important Papers” in the opinion of two referees. They will be published shortly (those marked with a diamond will be published in the next issue). Short summaries of these articles can be found on the *Angewandte Chemie* homepage at the address <http://www.angewandte.com>

A Nanoporous Metal–Organic Framework Based on Bulky Phosphane Ligands

X. Xu, M. Nieuwenhuyzen, S. L. James* ◆

Surface Structure and Crystal Growth in Zeolite Beta C

B. Slater,* R. A. Catlow, Z. Liu, T. Ohsuna, O. Terasaki, M. A. Camblor

Callipeltoside A: Assignment of Its Absolute and Relative Configuration by Total Synthesis

B. M. Trost,* O. Dirat, J. L. Gunzner ◆

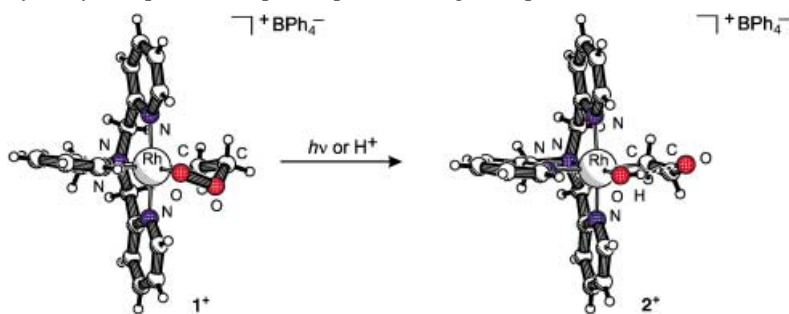
A New Catalyst for the Selective Oxidation of Butane and Propane

M. E. Davis,* C. J. Dillon, J. H. Holles, J. Labinger ◆

Surface Structure and Crystal Growth in Zeolite Beta C

B. Slater,* R. A. Catlow, Z. Liu, T. Ohsuna, O. Terasaki, M. A. Camblor

Instead of direct conversion into a metal oxo complex and acetaldehyde or ethylene oxide, as was proposed earlier for 3-metalla-1,2-dioxolanes, 3-rhoda-1,2-dioxolanes **1**⁺ have now been found to rearrange to rhodium formylmethyl hydroxy complexes **2**⁺ upon exposure to light or protons.



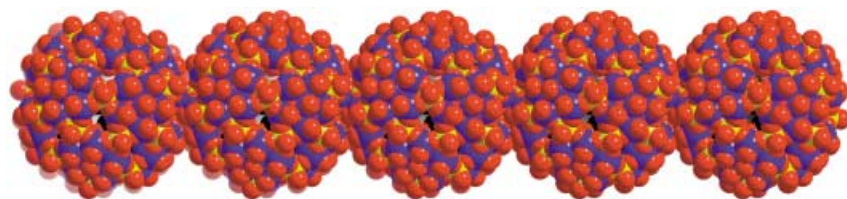
Angew. Chem. **2002**, *114*, 595–599

M. Krom, R. G. E. Coumans,
J. M. M. Smits, A. W. Gal* 575–579

Rearrangement of 3-Rhoda-1,2-dioxolanes to Rhodium Formylmethyl Hydroxy Complexes

Keywords: formylmethyl complexes • metallacycles • N ligands • photochemistry • rhodium

Crystal engineering with nanoobjects? Spherical structurally well-defined molybdenum-oxide-based giant clusters can be appropriately functionalized to a crystalline material with the remarkable property of having discrete cluster units which get covalently linked to form chains (see picture) through Fe–O–Mo bonds after having approached each other as a result of the release of crystal water upon drying.



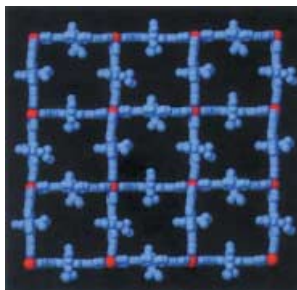
Angew. Chem. **2002**, *114*, 599–602

A. Müller,* S. K. Das,
M. O. Talismanova, H. Bögge,
P. Kögerler, M. Schmidtman,
S. S. Talismanov, M. Luban,
E. Krickemeyer 579–582

Paramagnetic Keplerate “Necklaces”
Synthesized by a Novel Room-Temperature Solid-State Reaction:
Controlled Linking of Metal-Oxide-Based Nanoparticles

Keywords: composites • magnetic properties • molybdenum • polyoxometalates • supramolecular chemistry

Use of a chiral, fluorene-based *N,N'*-bipyridine-type ligand (9,9-bis[(*S*)-2-methylbutyl]-2,7-bis-(4-pyridylethynyl)fluorene) with a Cu^{II} salt led to the formation of the first chiral non-interpenetrating square-grid coordination polymer (see picture). The free space in the polymer can be controlled by adjusting the size of the side chains attached to the ligand. The large channels and the incorporation of chirality into the polymer make such compounds strong candidates for chiral recognition applications.



Angew. Chem. **2002**, *114*, 603–605

N. G. Pschirer, D. M. Ciurtin,
M. D. Smith, U. H. F. Bunz,*
H.-C. zur Loye* 583–585

Noninterpenetrating Square-Grid Coordination Polymers With Dimensions of 25 × 25 Å² Prepared by Using *N,N'*-Type Ligands: The First Chiral Square-Grid Coordination Polymer

Keywords: chirality • coordination polymers • copper • square grids

Magnetic avalanches: A chiral polymeric manganese(II) complex with an enantiopure nitronyl nitroxide ligand (see structure of repeating unit; red: oxygen, green: nitrogen, blue: manganese) shows unusual magnetic behavior. The polymer orders magnetically in the bulk at 3 K, and reveals unusual dynamic and hysteretic magnetic effects as well as a remarkable magnetic avalanche phenomenon below 0.3 K.



Angew. Chem. **2002**, *114*, 606–609

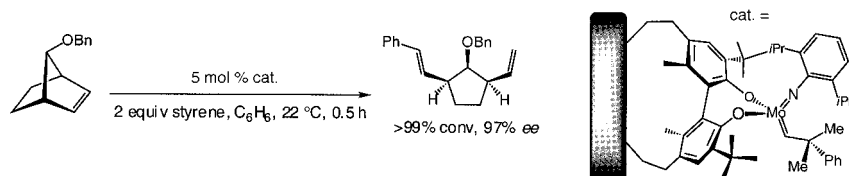
M. Minguet, D. Luneau,* E. Lhotel,
V. Villar, C. Paulsen, D. B. Amabilino,
J. Veciana* 586–589

An Enantiopure Molecular Ferromagnet

Keywords: chirality • ferromagnets • magnetic properties • manganese • O ligands • polymers



Substantially less toxic metal impurity than when unbound chiral complexes are used—this is achieved by the first polymer-supported chiral catalysts for olefin metathesis. These allow for efficient synthesis of various unsaturated carbo- and heterocycles in high optical purity through ring-opening (see scheme) and ring-closing reactions.



Angew. Chem. **2002**, *114*, 609–613

K. C. Hultsch, J. A. Jernelius,
A. H. Hoveyda,*
R. R. Schrock* 589–593

The First Polymer-Supported and
Recyclable Chiral Catalyst for
Enantioselective Olefin Metathesis

Keywords: asymmetric catalysis •
immobilization • metathesis •
molybdenum • solid-phase synthesis

A birefringent gel is formed when polycondensation of a strongly dipolar bis(trialkoxysilyl) compound is performed in an electric field. The birefringence is evidence for anisotropic ordering on the micrometer scale. The picture shows an image of the gel under a polarizing light microscope.



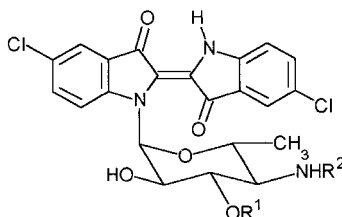
Angew. Chem. **2002**, *114*, 614–616

G. Cerveau, R. J. P. Corriu,* E. Framery,
S. Ghosh, M. Nobili 594–596

Nonrelaxable Anisotropic Organization
of Organic–Inorganic Hybrid Materials
Induced by an Electric Field

Keywords: birefringence • gels •
organic–inorganic hybrid composites •
self-assembly • sol–gel processes

Sky blue akashins (see picture) are the first indigoglycosides and the first natural derivatives of 5,5'-dichloroindigo to be isolated from terrestrial *Streptomyces*.



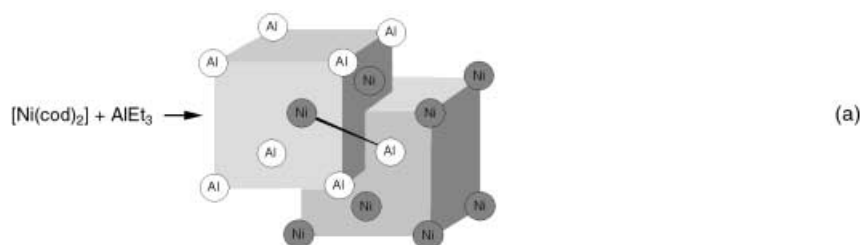
Angew. Chem. **2002**, *114*, 623–625

R. P. Maskey, I. Grün-Wollny,
H. H. Fiebig, H. Laatsch* 597–599

Akashins A, B, and C: Novel Chlorinated
Indigoglycosides from *Streptomyces* sp.
GW 48/1497

Keywords: antitumor agents •
glycosides • indigo • natural products

A fine powder of β -nickel aluminide NiAl is obtained by reaction of [Ni(cod)₂] (cod = cycloocta-1,5-diene) with AlEt₃ in toluene at room temperature and under a H₂ pressure of 5–10 MPa and subsequent distillation of the solvent, subsequent hydrogenation (0.1–5 MPa), and annealing (200 °C) [Eq. (a)]. This novel wet chemistry synthesis of aluminides may be transferred to Ni₃Al (α -nickel aluminide) and ternary aluminide powders such as GaNiAl, FeNiAl, and CeNiAl.



Angew. Chem. **2002**, *114*, 628–632

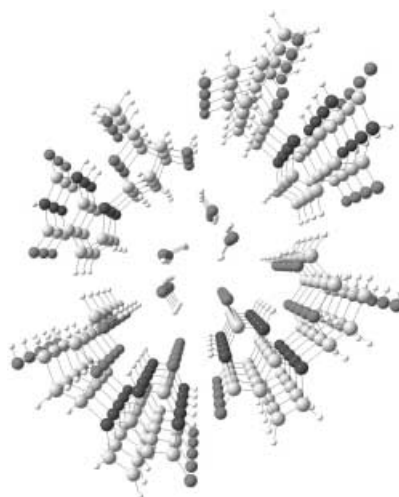
H. Bönemann,* W. Brijoux,
H.-W. Hofstadt, T. Ould-Ely, W. Schmidt,
B. Waßmuth, C. Weidenthaler . 599–603

Wet Chemistry Synthesis of β -Nickel
Aluminide NiAl

Keywords: aluminum • intermetallic
phases • metal–metal interactions •
nickel • solid-state structures



Supramolecular tetrameric aggregates of four molecules of 7-deaza-2'-deoxyxanthosine are stabilized by unconventional C–H...O hydrogen bonds. In the solid state the tetramers stack to form a nanotube (see picture), in the center of which water molecules are found which are also stacked.



Angew. Chem. **2002**, *114*, 617–619

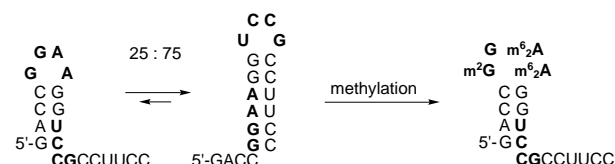
F. Seela,* T. Wiglenda, H. Rosemeyer,
H. Eickmeier, H. Reuter* 603–605

7-Deaza-2'-deoxyxanthosine Dihydrate
Forms Water-Filled Nanotubes with
C–H...O Hydrogen Bonds

Keywords: hydrogen bonds • nanotubes •
nitrogen heterocycles • nucleosides •
oligonucleotides



One sequence, but two conformations! The coexistence of conformers has been ascertained for a series of RNA oligonucleotides. The monomolecular conformation equilibria are shifted significantly by methylation of selected nucleobases; the methylation of the studied sequences corresponds to that of the naturally occurring helix 45 at the 3'-end in the small subunit of ribosomal RNA.



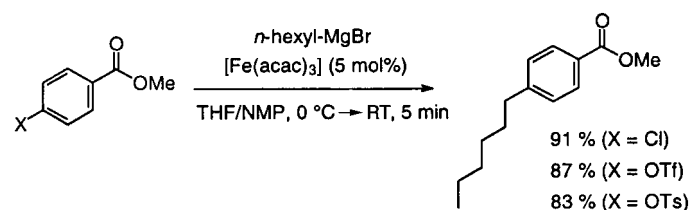
Angew. Chem. **2002**, *114*, 619–623

C. Höbartner, M.-O. Ebert, B. Jaun,
R. Micura* 605–609

RNA Two-State Conformation Equilibria
and the Effect of Nucleobase Methylation

Keywords: conformation analysis •
nucleosides • oligonucleotides •
RNA conformation equilibria •
RNA structures

Aryl chlorides are better substrates than the corresponding bromides or iodides in the presented cross-coupling with alkyl Grignard reagents that is catalyzed by iron salts (see scheme); even aryl tosylates are converted efficiently. This situation is attributed to a novel catalytic cycle, which probably involves iron complexes with formally negative oxidation states (probably Fe^{-II}).



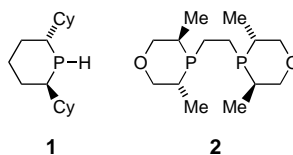
Angew. Chem. **2002**, *114*, 632–635

A. Fürstner,* A. Leitner 609–612

Iron-Catalyzed Cross-Coupling
Reactions of Alkyl-Grignard Reagents
with Aryl Chlorides, Tosylates, and
Triflates

Keywords: cross-coupling • Grignard
reagents • iron • magnesium •
organometallic compounds

Obvious but unknown in asymmetric catalysis were chiral six-membered-ring phosphanes and secondary phosphanes. As first examples of such ligands, oxaphosphinanes were now prepared and examined in asymmetric hydrogenation. With the monodentate oxaphosphinane **1**, for example, 96% *ee* was achieved with itaconic acid as the substrate and 97.5% *ee* was achieved with the chelate ligand **2** and 2-acetamidoacrylic acid as the substrate.



Angew. Chem. **2002**, *114*, 625–628

M. Ostermeier, J. Prieß,
G. Helmchen* 612–614

Mono- and Bidentate Phosphinanes—
New Chiral Ligands and Their
Application in Catalytic Asymmetric
Hydrogenations

Keywords: asymmetric catalysis • chelate
ligands • hydrogenation • phosphane
ligands • P ligands

Complementary strands and redox intercalators can be self-assembled into nanoscale structures capable of charge transfer at the electrode surface from DNA oligomers synthesized directly at covalently modified semiconductor silicon surfaces (see schematic representation).



A. R. Pike, L. H. Lie, R. A. Eagling, L. C. Ryder, S. N. Patole, B. A. Connolly, B. R. Horrocks, A. Houlton* .. 615–617

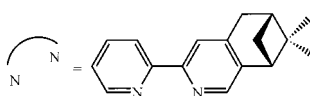
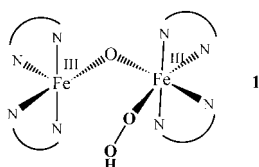
DNA On Silicon Devices: On-Chip Synthesis, Hybridization, and Charge Transfer

Keywords: charge transfer • DNA • molecular devices • semiconductors • silicon

Angew. Chem. **2002**, *114*, 637–639



In a spin: Transient peroxo iron complexes have been characterized by the means of Mössbauer and HF-EPR spectroscopies. Complex **1** has an unexpected $S=2$ ground state resulting from the coupling between the $S=5/2$ and $S=1/2$ iron(III) ions.



H. Hummel, Y. Mekmouche, C. Duboc-Toia, R. Y. N. Ho, L. Que, Jr., V. Schünemann, F. Thomas, A. X. Trautwein, C. Lebrun, M. Fontecave,* S. Ménage* ... 617–620

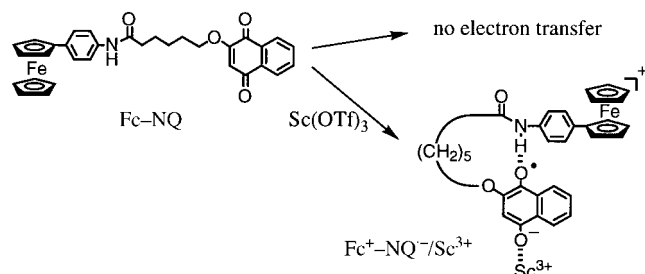
A Diferric Peroxo Complex with an Unprecedented Spin Configuration: An $S=2$ System Arising from an $S=5/2$, $1/2$ Pair

Keywords: EPR spectroscopy • iron • metalloenzymes • O–O activation • peroxo ligands

Angew. Chem. **2002**, *114*, 639–642



Initiated by metal ions: Intramolecular electron transfer in a donor–acceptor linked system (a ferrocene–naphthoquinone dyad Fc–NQ; see scheme) can be started thermally by adding an appropriate metal ion (Sc^{3+}) which can promote thermal electron transfer, which would otherwise be unlikely to occur, to give a radical ion complex ($\text{Fc}^+-\text{NQ}^-/\text{Sc}^{3+}$).



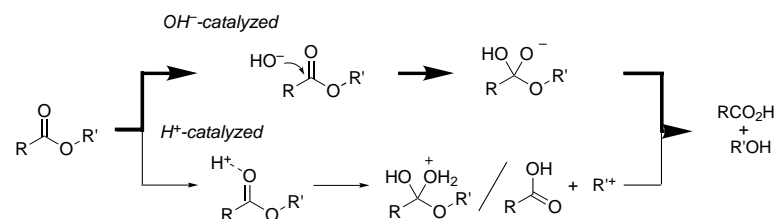
S. Fukuzumi,* K. Okamoto, H. Imahori* 620–622

Thermal Intramolecular Electron Transfer in a Ferrocene–Naphthoquinone Linked Dyad Promoted by Metal Ions

Keywords: electron transfer • ESR • ferrocene • kinetics • scandium

Angew. Chem. **2002**, *114*, 642–644

Not protons but hydroxide ions catalyze ester hydrolysis in supercritical water when the concentration of carboxylic acid generated by hydrolysis is low, as in the initial stage of the reaction. As the concentration of carboxylic acid increases with the progress of the reaction, the mechanism would become proton catalyzed (see scheme).



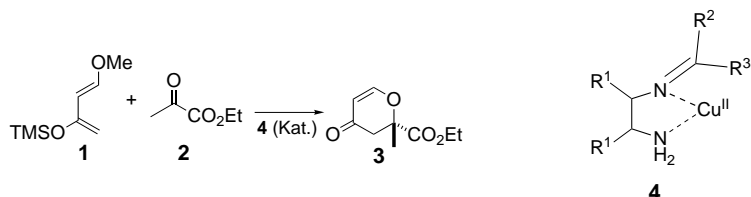
H. Oka, S. Yamago,* J. Yoshida,* O. Kajimoto* 623–625

Evidence for a Hydroxide Ion Catalyzed Pathway in Ester Hydrolysis in Supercritical Water

Keywords: esters • hydrolysis • kinetics • supercritical fluids

Angew. Chem. **2002**, *114*, 645–647

Only the correct order of addition of the components is required to prepare a variety of chiral Lewis acids. The complexes (for example, **4**) formed by the condensation of chiral, nonracemic 1,2-diamines with ketones and subsequent addition of $\text{Cu}(\text{OTf})_2$ (Tf = trifluoromethanesulfonyl) catalyze the cycloaddition reaction of Danishefsky's diene (**1**) and ethyl pyruvate (**2**) to form dihydropyrone (**3**).



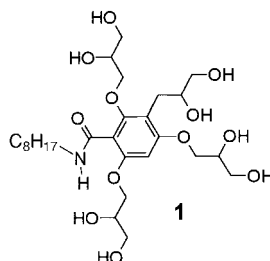
Angew. Chem. **2002**, *114*, 647–650

P. I. Dalko,* L. Moisan,
J. Cossy* 625–628

Modular Ligands for Asymmetric
Synthesis: Enantioselective Catalytic
 Cu^{II} -Mediated Condensation Reaction of
Ethyl Pyruvate with Danishefsky's Diene

Keywords: asymmetric catalysis • chiral
auxiliaries • cycloaddition • Lewis acids

A new mesophase, composed of closed spheroidal aggregates that contain lipophilic alkyl chains within a continuum of hydrogen-bonding networks, was realized with the low molecular weight amphiphile **1** in the absence of water or any other solvent.



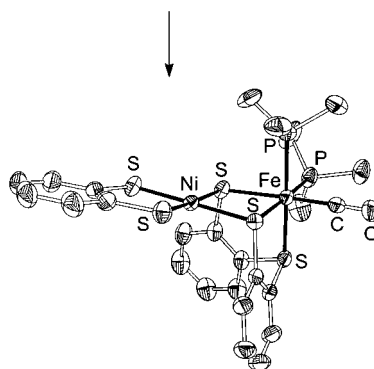
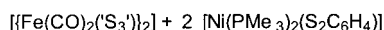
Angew. Chem. **2002**, *114*, 650–653

P. Fuchs, C. Tschierske,* K. Raith, K. Das,
S. Diele 628–631

A Thermotropic Mesophase Comprised of
Closed Micellar Aggregates of the
Normal Type

Keywords: amphiphiles • hydrogen
bonds • liquid crystals • mesophases •
micelles

Attempts to model the active sites of $[\text{NiFe}]$ hydrogenases have yielded numerous NiFe thiolate complexes but never a species with the characteristic $[(\text{RS})_2\text{Ni}(\mu\text{-SR})_2\text{Fe}(\text{CO})(\text{CN})_2]$ core. The $\text{Ni}^{\text{II}}/\text{Fe}^{\text{II}}$ title complex represents the PMe_3 derivative. Its $\nu(\text{CO})$ frequency of 1948 cm^{-1} corresponds to that of 1940 cm^{-1} for $[\text{NiFe}]$ hydrogenase in the Ni-R state.

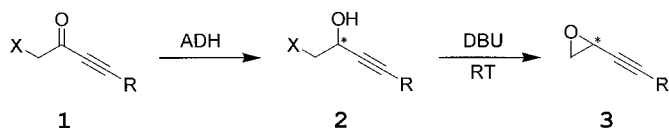


D. Sellmann,* F. Geipel, F. Lauderbach,
F. W. Heinemann 632–634

$[(\text{C}_6\text{H}_4\text{S}_2)\text{Ni}(\mu\text{-S}_3')\text{Fe}(\text{CO})(\text{PMe}_3)_2]$: A
Dinuclear $[\text{NiFe}]$ Complex Modeling the
 $[(\text{RS})_2\text{Ni}(\mu\text{-SR})_2\text{Fe}(\text{CO})(\text{L})_2]$ Core of
 $[\text{NiFe}]$ Hydrogenase Centers

Keywords: carbonyl ligands •
hydrogenases • iron • nickel • S ligands

Halogenated propargylic alcohol 2a ($\text{X} = \text{Cl}$, $\text{R} = \text{Ph}$) was prepared by the enzymatic reduction of the corresponding ketone **1a** with excellent enantioselectivity and chemical yield. Both enantiomers were synthesized with high total turnover numbers of the catalysts and cofactors. Other halogenated alkynones **1** were also suitable substrates for the enzymes. Alcohols **2** were converted into propargylic epoxides **3** under mild conditions to give multifunctionalized building blocks. $\text{X} = \text{Cl}$, Br ; $\text{R} = \text{H}$, Ph , trimethylsilyl, *tert*-butyldimethylsilyl; ADH = alcohol dehydrogenase; DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene.



Angew. Chem. **2002**, *114*, 656–659

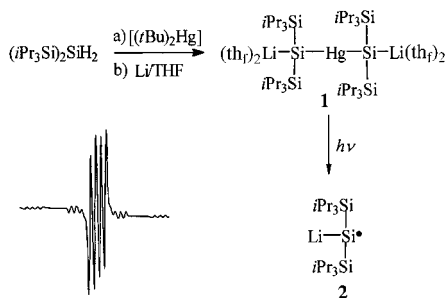
T. Schubert, W. Hummel,
M. Müller* 634–637

Highly Enantioselective Preparation of
Multifunctionalized Propargylic Building
Blocks

Keywords: asymmetric catalysis •
enzyme catalysis • epoxides • ketones •
oxidoreductases • reduction



Radical developments in silyllithium chemistry: The preparation and X-ray structure of **1**, the first compound with geminal Hg-Si-Li bonding, is reported. Compound **1** is obtained by the reaction of $(i\text{Pr}_3\text{Si})_2\text{SiH}_2$ with $[(t\text{Bu})_2\text{Hg}]$ followed by lithiation in THF. Irradiation of **1** produces $[\text{Li}(i\text{Pr}_3\text{Si})_2\text{Si}]^\bullet$ (**2**), the first silyl radical with an α -Si-Li bond detected by EPR spectroscopy (see spectrum).



D. Bravo-Zhivotovskii,* M. Yuzevovich, N. Sigal, G. Korogodsky, K. Klinkhammer, B. Tumanskii, A. Shames, Y. Apeloig* 649–651

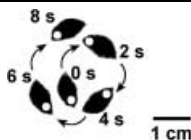
The Synthesis of the First Compound with Li-Si-Hg Bonding: $[\text{Li}(i\text{Pr}_3\text{Si})_2\text{Si}]\text{Hg}$ —a Source for the $[\text{Li}(i\text{Pr}_3\text{Si})_2\text{Si}]^\bullet$ Radical

Keywords: EPR spectroscopy • lithium • mercury • radicals • silicon

Angew. Chem. 2002, 114, 671–673



The artificial millimeter-scale “autonomous movers” glide across the surface of a liquid without an external power source. This system is based on a combination of two processes: Motion of individual objects powered by the catalytic decomposition of hydrogen peroxide, and relative motion (self-assembly) caused by capillary interactions at the fluid/air interface. The picture shows the rotational/translational motion of a single object; the motion of a pair of these object depends on their chirality.



R. F. Ismagilov, A. Schwartz, N. Bowden, G. M. Whitesides* 652–654

Autonomous Movement and Self-Assembly

Keywords: autonomous movement • complexity • energy-dissipating systems • self-assembly

Angew. Chem. 2002, 114, 674–676



Supporting information on the WWW (see article for access details).

* Author to whom correspondence should be addressed



Accelerated publications



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Southampton
Electrochemistry Group

R. Holze 655

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Michael V. Mirkin

R. Schuster 657

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R. Eric Banks

T. M. Klapötke 658

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Hubert Girault

C. Amatore 658

Dictionary of Common Names/Trivialnamen-Handbuch

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Chemie

K.-H. Hellwich 659

Design of Molecular Materials

Jacques Simon,
Pierre Bassoul

H. Frey 660



WEB SITES

<http://www.ill.fr/dif/3D-crystals/>

Making Matter—The Atomic Structure of Materials

W. Milius 661

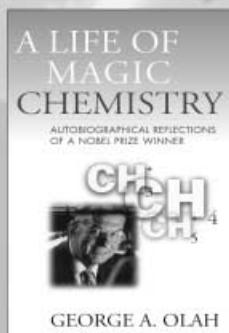
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Issue 3, 2002 was published online on January 29.

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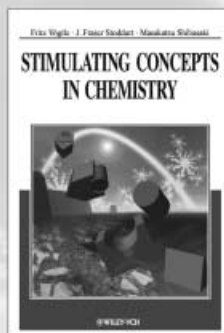
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